



Comparative profiles of *Achyrocline alata* (Kunth) DC. and *A. satureioides* (Lam.) DC., Asteraceae, applying HPLC-DAD-MS

Rafaela Grassi-Zampieron,¹ Leonardo V. França,² Carlos A. Carollo,² Maria do Carmo Vieira,³ Alberto Oliveros-Bastidas,⁴ João M. de Siqueira^{*5}

¹Curso de Pós-graduação Interinstitucional em Ciências da Saúde, UFMS/UFMG/UNB, Brazil, Caixa Postal 549, 79070-900 Campo Grande-MS, Brazil,

²Departamento de Farmácia-Bioquímica, Centro de Ciências Biológicas e da Saúde, Universidade Federal de Mato Grosso do Sul, Caixa Postal 549, 79070-900 Campo Grande-MS, Brazil,

³Faculdade de Ciências Agrárias, Universidade Federal da Grande Dourados, Caixa Postal, 533, 79804-970 Dourados-MS, Brazil,

⁴Universidad de los Andes, Facultad de Ciencias, Depto de Química, Grupo de Química Ecológica, Nucleo Universitario Pedro Rincón Gutiérrez, 5101-A, Mérida, Venezuela,

⁵Laboratório de Farmacognosia/Produtos Naturais, Campus Centro-Oeste Dona Lindu, Universidade Federal de São João Del Rei, Rua Sebastião Coelho, 400, 35501-296 Divinópolis-MG, Brazil.

RESUMO: “Perfil comparativo entre *Achyrocline alata* (Kunth) DC. e *A. satureioides* (Lam.) DC., Asteraceae, utilizando HPLC-DAD-MS”. Análises comparativas entre os extratos hidrometanólico de *Achyrocline alata* (Kunth) DC. e *A. satureioides* (Lam.) DC., Asteraceae, foram desenvolvidas utilizando HPLC-DAD-MS. Ambas as plantas são utilizadas indistintamente para as mesmas indicações na medicina popular de Mato Grosso do Sul, Brasil, enquanto a *A. alata* (“jatei-ka-há”) é predominante neste estado, a *A. satureioides* (“marcela”) é predominante nos demais estados da Federação. Amostras das duas espécies coletadas em épocas diferentes, 1996 e 2002, apresentaram o mesmo perfil cromatográfico. O presente resultado pode justificar que *A. alata* possa ser utilizada como sucedânea de *A. satureioides*, pois apresentam similar composição de metabólitos secundários polares.

Unitermos: *Achyrocline*, chalcona, cafeoil, perfil cromatográfico, HPLC-DAD-MS.

ABSTRACT: A comparative analysis between the hydromethanolic extracts of *Achyrocline alata* (Kunth) DC. and *A. satureioides* (Lam.) DC., Asteraceae, was performed by the use of HPLC-DAD-MS. Both plants were used without distinction and under the same indications in folk medicine in Mato Grosso do Sul, Brasil. While *Achyrocline alata* (“jatei-ka-ha”) is used in folk medicine of the Brazilian state of Mato Grosso do Sul, *A. satureioides* is predominantly used in other states. Samples of both plants collected in different periods, 1996 and 2002, showed a very similar chemical profile. The results indicate that *A. alata* could be used in phytotherapeutic preparations as substitute for *A. satureioides*, since they have similar chemical compositions of the polar extract.

Keywords: *Achyrocline*, chalcone, caffeoyl derivatives, chromatographic profile, HPLC-DAD-MS analysis.

INTRODUCTION

The *Achyrocline* genus belongs to the Asteraceae family and comprises from twenty to thirty species spread through tropical and subtropical regions of South and Central Americas, Africa and Madagascar (Bremer, 1994). Asteraceae has been one of the most studied plant families, several species are widely used in South America, such as: *A. alata*, *A. tomentosa*, *A. flaccida* and *A. satureioides*. Previous chemical investigations of *A. alata* and other species of *Achyrocline* collected in South

America (Argentina, Uruguay) were performed, showing similar profiles regarding to their phenolic constituents, flavonoids and quinic acid derivatives (Lopez et al., 2006, Broussalis et al., 1993).

In Brazil, *A. satureioides* (Lam.) DC. (popularly called “marcela”) is predominant in folk medicine in most of Brazilian states, while *A. alata* (known as “jatei-ka-ha”) is adopted in Mato Grosso do Sul state as a substitute of the former. Its yellow inflorescence is slightly more yellow besides having a more pronounced odour when compared to *A. satureioides*. A survey of the medicinal

*E-mail: jmaximo@ufsj.edu.br; Tel. +55 37 3221 1392.

plants more requested by clients or indicated by herb sellers operating in the central area of Campo Grande-MS, Brazil, was performed in two different times - 1992 and 2002. In both surveys, "jatei-ka-ha" was the third plant among the most requested ones and/or indicated by them, while "marcela" was fifteenth and nineteenth, respectively. Both plants have been used under the same preparation procedure, as well as with the same indications in folk medicine (Nunes et al., 2003). In others surveys, made in some cities of the state, in special those towards the Brazilian border with the Latin American country, in this case Paraguay and Bolivia, *A. alata* continued being cited as the most used species in popular medicine (Siqueira & Palhano, 1994). However, no pharmacological report of *A. alata* was found in the literature, whereas previously pharmacological evaluations were reported about *A. saturoioides*, which showed antispasmodic and relaxing effects; anti-inflammatory (Simões et al., 1988, Hnatyszyn et al., 2004, Fachinnetto et al., 2007), analgesic and sedative activities, effects on intestinal transit and antioxidant action (Desmarchelier et al., 1998, Leal et al., 2006, Grassi-Zampieron et al., 2009).

Therefore, the present study was carried out to compare the chromatogram profiles obtained by HPLC-DAD-MS of these both species cultivated in different times - 1996 and 2002. In addition, a further phytochemical workup of the ethyl acetate fraction obtained from hydromethanolic extract from *A. alata* was performed.

MATERIAL AND METHODS

Plant material

Inflorescences of *Achyrocline alata* (Kunth) DC. and *A. saturoioides* (Lam.) DC., Asteraceae, were cultivated and collected in the Campus of the Universidade Federal da Grande Dourados (UFGD), Dourados-MS, Brazil, on April of the 1996 and in 2002, and identified by Dr. Lilian Auler Mentz. Voucher specimens of the plants were kept in the CG/MS herbarium (numbers 11486 and 11487, respectively).

Climatic and horticulture conditions

The plants were cultivated on April of 1996 and 2002 in the Horto de Plantas Mediciniais (HPM), of the UFGD, in Dourados. Dourados city has medium altitude of 452 m and its climate is classified by the international system of Köppen as Cwa - humid subtropical climate (Mato Grosso do Sul State in 1990). The annual medium precipitation is nearly 1500 mm and the annual medium temperature is about 22 °C. The average relative humidity of the air during this period was of 73.18% (Estação Agroclimatológica - UFMS). The soil is originally cerrado vegetation, and it is classified as typical Haplortox of loamy texture and flat topography. Chemical analyses of

samples of the soil showed: pH:H₂O (1:2.5) = 5.3 and 6.0; P (mg/dm³) = 14.0; K (μmol/dm³) = 3.3; Al³⁺ (μM/dm³) = 0.0; Ca²⁺ (μM/dm³) = 30.0; Mg (μM/dm³) = 14.9; H + Al (μM/dm³) = 78.6; Sb (μM/dm³) = 42.9; CTC (μMdm³) = 121.5; V (%) = 33.0 and Mo (g/kg) = 29,6.

Extracts obtained from *A. alata* and *A. saturoioides* for chromatography evaluations by LC-MS analysis and usual work up.

Extracts for liquid chromatography, instrumentation and conditions

Extracts for HPLC-DAD-MS analysis were obtained by the following procedures: dried and powdered inflorescences of *A. alata* and *A. saturoioides* (100 mg) were extracted with 3 mL of MeOH:H₂O (7:3) and 3 mL of methylene chloride was also additionated with the aim of removing apolar compounds from hydromethanolic extract. Then, this mixture was submitted to ultrasonic extraction for 20 minutes followed by centrifugation and concentration; 20 μL of hydromethanolic was used. The HPLC-DAD-MS analysis were performed on a Shimadzu LC-20A HPLC apparatus with a diode array detector (CBM20A; Shimadzu) coupled to an UltraTOFq (Bruker Daltonics) ESI-qTOF mass spectrometer. For the analytical HPLC method a monolithic column (Onyx™ 150 x 4,6 mm - C-18 - Phenomenex) and a pre-column of the same material were used. The mobile phase consisted of a linear gradient elution at a flow rate of 3 mL/min with the following mixtures: water:acetic acid 1% (v/v) (phase A) and acetonitrile:acetic acid 1% (phase B) as eluents. The first 8 min were eluted with B phase with gradient 5-12%, following 13-22% (8-25 min), 22-35% (25-29 min), 35% (29-30 min), 35-100% (30-34 min). The standards used were obtained from the following sources: chlorogenic acid (Acros) (**1**, 4.28 min); isoquercetrin (Merck) (**2**, 11.98 min.) and quercetin (Merck) (**8**, 23.45 min.). 4,2',4'-Trihydroxy-6'-methoxychalcone (**10**, 30.77 min.), gnaphalin (**11**, 32.32 min.), 3-O-methyl-quercetin (**9**, 26.23 min.) were isolated from *A. alata*. 3,5 Dicafeoyl-quinic acid (**4**, 15.35 min.) and 4,5 dicafeoyl-quinic acid (**6**, 17.78 min.) were kindly given by Prof. Dr. Norberto Peoporine Lopes. The retention time and UV spectra of these standards are shown in Table 1 and Figure 1.

Extract from *A. alata* for fractionation on chromatography column

The hydromethanolic extract from dried inflorescences of *A. alata* (100 g) was obtained using MeOH:H₂O (7:3) at 40 °C (19.2 g) and was concentrated under reduced pressure. A sample of hydromethanolic extract from *A. alata* (6 g) was subjected to liquid-liquid extraction, yielding hexane (0.9 g) and ethyl acetate (1.5 g) fraction. The fractionation of the ethyl acetate fraction on

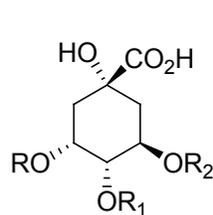
silica gel column using CHCl₃:MeOH gradient, furnished 4,2',4'-trihydroxy-6'-methoxychalcone (**10**), besides other flavonoids already isolated previously (3-*O*-methylquercetin **9**, quercetin **8**, and gnapthalin **11**). This chalcone was determined by 1D- and 2D-NMR spectrometry by a comparison of some models available in the literature (Morikawa et al., 2009, Puyvelde et al., 1989, Slimestad et al., 2008), and by UV and ESI-MS data. Spectral data of the other known flavonoids (UV, NMR) were also compared with those described in the literature (Agrawal et al., 1989).

4,2',4'-Trihydroxy-6'-methoxychalcone. UV_{max} (MeOH): 230 and 340 nm; ¹H-NMR (400 MHz, DMSO-d₆): δ 13.94 (br s, H), 10.87 (br s, H), 10.10 (br s, H), disappeared after addition of D₂O, 7.66 (d, *J* = 15.1 Hz, H-α), 7.58 (d, *J* = 15.1 Hz, H-β), 7.36 (d, *J* = 8.4 Hz, H-2 and 6), 6.76 (d, *J* = 8.4 Hz, H-3 and 5), 5.88 (d, *J* = 2.0 Hz, H-5'), 5.86 (d, *J* = 2.0 Hz, H-3'), 3.82 (s, OCH₃ in C-6'); ¹³C-NMR (100 MHz, CDCl₃/DMSO-d₆): 104.6 (C-1'), 167.0 (C-2'), 90.8 (C-3'), 164.3 (C-4'), 95.8 (C-5'), 162.3 (C-6'), 55.1 (OCH₃ in C-6'), 191.3 (C=O, C-β'), 123.4 (C-α), 141.9 (C-β), 125.9 (C-1), 129.5 (C-2 and C-6), 115.4 (C3 and C-5), 159.2 (C-6). Main HMBC correlations observed: δH/δC 3.82(OCH₃)/162.3 (C-6'), 5.88(H-5')/90.8(C-3'), 5.86(H-3')/95.8(C-5'), 7.66(H-α)/191.3(C-β'), 7.58(H-β)/191.3(C-β'); by ESI-MS in positive mode *m/z* 287 [M+H]⁺; *m/z* 167 [C₈H₇O₄]⁺ and in negative mode *m/z* 285 were observed.

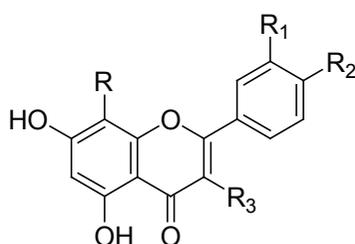
RESULTS AND DISCUSSION

The plants showed phenylpropanoids and flavonoids derivatives as the major compounds, initially

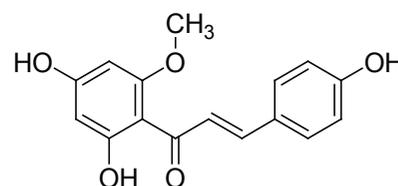
characterized by their UV spectra (Table 1 and Figure 1). Among these, three dicaffeoylquinic acid derivatives were observed (peaks 3, 4 and 6); the peaks 4 and 6 were identified by comparison with authentic standards as 3,5- and 4,5-dicaffeoylquinic acids, respectively. The peak 3 (retention time 14.60 min) was analysed and its fragments were compared with the identification keys of these compounds proposed by Clifford et al., (2003; 2005). In these identification keys the fragment *m/z* 173 [M-2caf-H₂O]⁻ is characteristic of derivatives substituted in the position 4, besides the absence of the ions characteristic of 1,4-dicaffeoylquinic acid (*m/z* 299 and *m/z* 203), suggests the identification of the peak 3 as being 3,4-dicaffeoylquinic acid. The peak 5 presented a UV spectrum similar to the one observed for the flavonoid quercetin; this compound showed an ion in *m/z* 463.0846, compatible with the chemical formula C₂₁H₁₉O₁₂⁻ (463.0876). This formula is the same presented by the flavonoid isoquercetrin (**2**), besides that, it was observed in the MS-MS spectrum (collision energy 15 eV) as main fragment the ion *m/z* 301, compatible with aglycone quercetin, this fragment suggested the loss of hexosyl residues (*m/z* 162). By analysing these data is possible to characterize this compound as a flavonol quercetin derivative, probably glucosilated or galactosilated. The peak 7 showed a UV spectrum similar to the compound previously described, however showing a molecular weight of *m/z* 447.1259. In the MS-MS spectra (collision energy 15 eV) it was observed that the main fragment is the ion *m/z* 285, compatible with the flavonol kaempferol, so with the presented data and by analogy with the compound previously described it is possible to characterize this peak as the flavonol kaempferol derivative probably glucosilated or galactosilated. The other picks were characterized by comparing standards by



- 1** R = R₁ = H; R₂ = X
3 R₂ = R; R = R₁ = X
4 R₁ = H; R₁ = R₂ = X
6 R = H; R₁ = R₂ = X



- 2** R = H; R₁ = R₂ = OH; R₃ = glu
8 R = H; R₁ = R₂ = R₃ = OH
9 R = H; R₁ = R₂ = OH; R₃ = OMe
11 R = OMe; R₁ = R₂ = H; R₃ = OMe



10

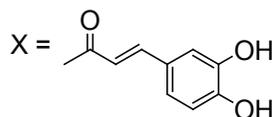
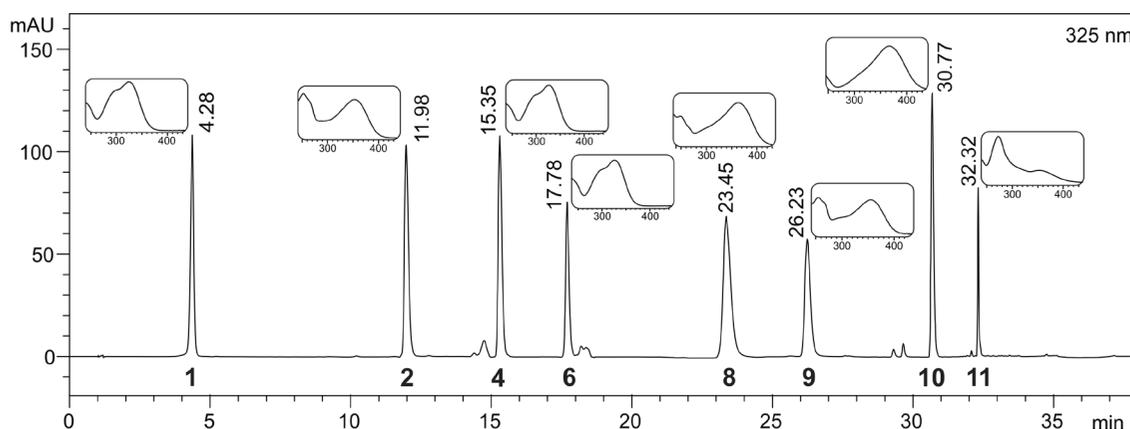


Table 1. Identification of main compounds in the hydromethanolic extracts from *A. alata* and *A. satureioides* by retention time, UV spectra and negative ions in HPLC-DAD-ESI-MS.

Peaks N ^o	tR (min)	Compound	UV max (nm)	Negative ionization MS (m/z)	Chemical Formula
1	4.28	Chlorogenic acid	299, 325	353.0877 [M-H] ⁻	C ₁₆ H ₁₇ O ₉ ⁻
2	11.98	Isoquercetrin	254, 267 ^{sh} , 361	463.0852 [M-H] ⁻	C ₂₁ H ₁₉ O ₁₂ ⁻
3	14.60	3,4-Dicaffeoyl quinic acid	300, 325	515.1150 [M-H] ⁻	C ₂₅ H ₂₃ O ₁₂ ⁻
4	15.35	3,5-Dicaffeoyl quinic acid	299, 325	515.1150 [M-H] ⁻	C ₂₅ H ₂₃ O ₁₂ ⁻
5	16.30	Unidentified quercetin derivative	252, 267 ^{sh} , 364	463.0846 [M-H] ⁻	C ₂₁ H ₁₉ O ₁₂ ⁻
6	17.78	4,5-Dicaffeoyl quinic acid	299, 324	515.1146 [M-H] ⁻	C ₂₅ H ₂₃ O ₁₂ ⁻
7	22.53	Unidentified kaempferol derivative	229, 358	447.1259 [M-H] ⁻	C ₂₁ H ₁₉ O ₁₁ ⁻
8	23.45	Quercetin	254, 266 ^{sh} , 364	301.0327 [M-H] ⁻	C ₁₅ H ₉ O ₇ ⁻
9	26.23	3-O-Methyl-quercetin	255, 267 ^{sh} , 355	315.0491 [M-H] ⁻	C ₁₆ H ₁₁ O ₇ ⁻
10	30.77	4,2',4'-Trihydroxy-6'-methoxychalcone	367	285.0756 [M-H] ⁻	C ₁₆ H ₁₃ O ₅ ⁻
11	32.32	Gnaphaliin	265, 314 ^m	313.0710 [M-H] ⁻	C ₁₇ H ₁₃ O ₆ ⁻

sh - shoulder, m minor peak

**Figure 1.** Chromatogram of the standards used for identification of the compounds present in the two species. In detail can be observed the UV spectra of these compounds (250-450 nm).

retention time, UV spectra and molecular weight in high resolution.

The HPLC analysis with diode array detection revealed similar composition of the two species (Figure 2), without a great variation for the different years when the species were collected. This similarity was also confirmed by comparison of the spectra generated by the mass analysis (data is not presented). It can only be observed a quantitative variation between the species, mainly for the chalcone 4,2',4'-trihydroxy-6'-methoxychalcone, that shows a larger accumulation in *A. alata*. By usual phytochemical work up, this chalcone was isolated from *A. alata* along with flavonoids already isolated from this species, previously, but it is being described for the first time in these species.

Determination of caffeoyl derivatives content in native *Achyrocline* species from Argentina and Uruguay were recently evaluated, being observed a similar chemical profile between them. According to them, the major

content of caffeoyl derivatives in species of *Achyrocline* could justify some of their uses in folk medicine (Lopez et al., 2006). Finally, due to great chemical similarity among the secondary metabolites of the two species could be suitable justify for the indication of the *A. alata* to be used in phytotherapeutic preparations as a substitute for *A. satureioides* in the folk medicine, despite further pharmacological evaluations are necessary to confirm this point of view.

ACKNOWLEDGMENTS

The authors are grateful to FUNDECT-MS, Brazil, for the financial support and to PIBIC-CPq-PROPP-UFMS for a fellowship. Thanks are also given to FCFRP-USP-SP and Professor Norberto Peporine Lopes for making their facilities available for additional analysis (HPLC-DAD-MS), and the Chemistry Department of UFMG to obtain the NMR spectral data.

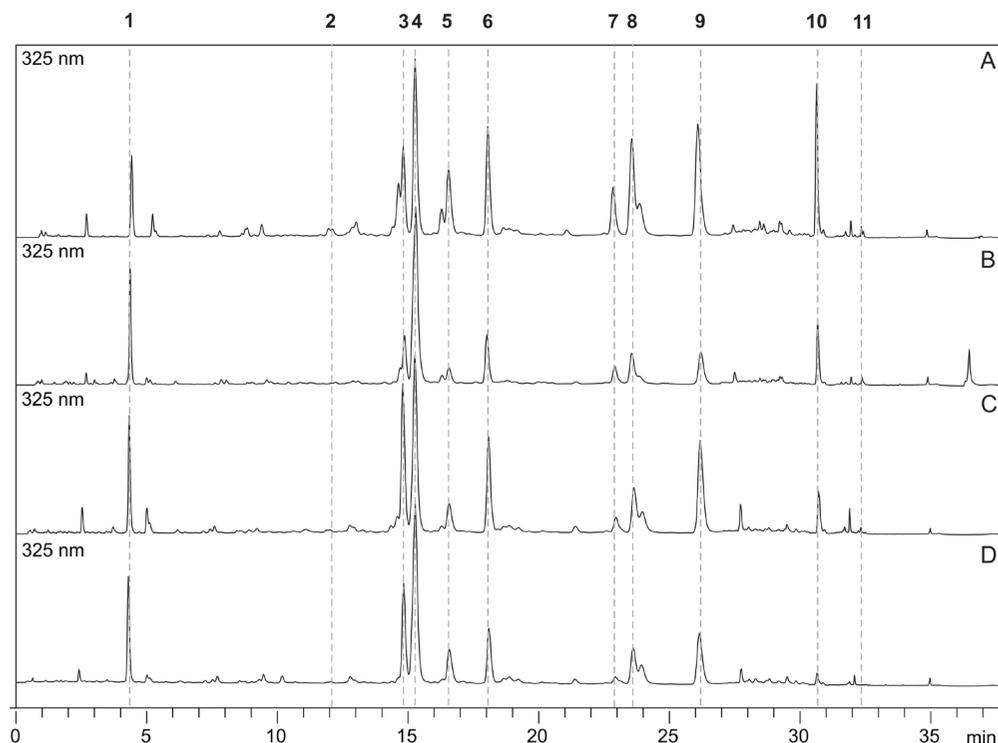


Figure 2. Chromatograms of the two species *A. alata* (A. collected in 1996; B. collected in 2002) and *A. satureioides* (C. collected in 1996; D. collected in 2002). Detection at 325 nm. The chromatographic peak identities are reported in Table 1.

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